ISO 21457 Materials Selection And Corrosion Control For Oil And Pdf

	A	B	C	D	E
1	**Last Name**	**Sales**	**Country**	**Quarter**	
2	Smith	$16,753.00	UK	Qtr 3	
3	Johnson	$14,808.00	USA	Qtr 4	
4	Williams	$10,644.00	UK	Qtr 2	
5	Jones	$1,390.00	USA	Qtr 3	
6	Brown	$4,865.00	USA	Qtr 4	
7	Williams	$12,438.00	UK	Qtr 1	
8	Johnson	$9,339.00	UK	Qtr 2	
9	Smith	$18,919.00	USA	Qtr 3	
10	Jones	$9,213.00	USA	Qtr 4	
11	Jones	$7,433.00	UK	Qtr 1	
12	Brown	$3,255.00	USA	Qtr 2	
13	Williams	$14,867.00	USA	Qtr 3	
14	Williams	$19,302.00	UK	Qtr 4	
15	Smith	$9,698.00	USA	Qtr 1	
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When selecting or developing a corrosion protection system, the following general rules should be applied: 1. The corrosion protection system shall be evaluated based on the systems . 2. The . 3. The corroding medium is considered. 4. The material to be protected shall be considered. 5. The operating conditions shall be considered. 6. The material selection shall be based on the condition. 7. The conditions shall be considered for . 8. The materials, design, and evaluation of systems shall be addressed in a structured and reproducible manner. Introduction ========== This section describes the condition of materials and the corresponding action of the materials. The purpose of the first of these is to draw attention to the corrosion potential of the material. The second is to take an inventory of the corrosion potentials of materials of interest in the context of the present work. The following sections explain the selected materials in the context of their potential and the corresponding protection measures. The actual state of a material is described by the . The first part of this defines the state, which is a quantitative and qualitative characterization of the . The second part of this defines a process and is a more qualitative way to describe the . This is a quantification of the state of the material as a function of time. The defines the corrosion potential of a material. The is defined as the difference between the and the of an electrolyte. The defines the . The is an indication of the potential of the material and can be measured using a . The state of the material and, therefore, its corrosion potential is defined by the . The defines the of a material. The defines the of a material, which can be for as well as. The is an indication of the potential of the material. For a material, the indicates the potential of the material and can be for as well as. The is a function of time. It is not absolute, since it is relative to the . In order to avoid confusion, the will be defined relative to the , and the relative to the . The defines the in the at a given . A 82157476af